

Sustainable Steel?

Christina Chang
Harvard Energy Journal Club
2/14/2019



The scale of the materials challenge

Definitive anthropogenic CO₂ emissions contribute significantly to total CO₂ emissions

Industry is greatest single sector for CO₂e

Just 5 materials account for 55% of industrial emissions

Figure 2.1—Pie charts showing the sources of global CO₂ emissions

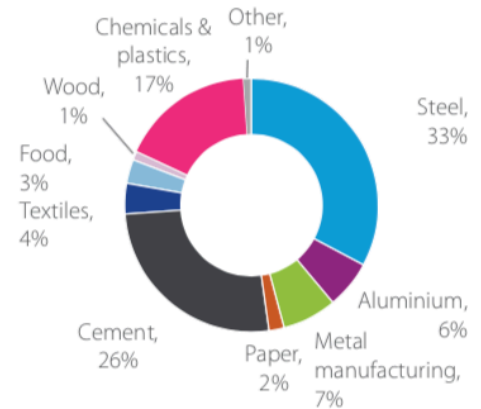
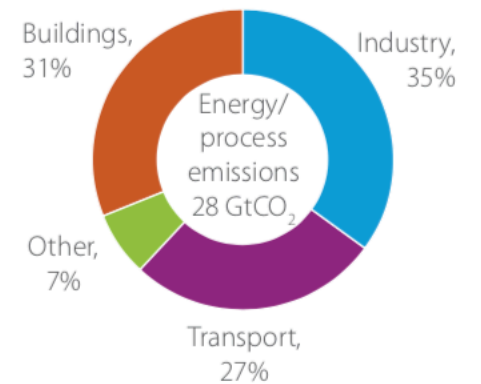
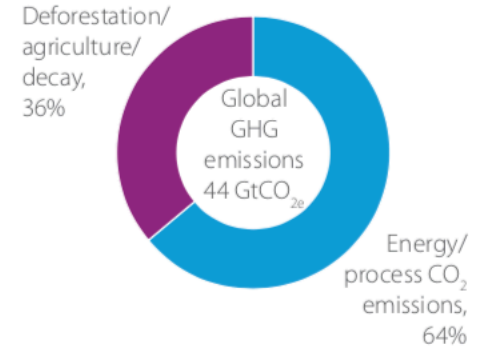
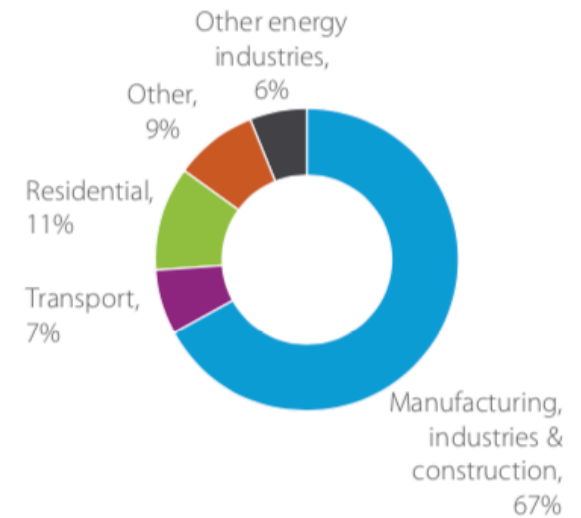


Figure 2.4—Sources of Chinese CO₂ emissions

A more detailed breakdown (China) confirms that stock production is largest CO₂e source

China's CO₂e profile



Manufacturing, industries, & construction breakdown

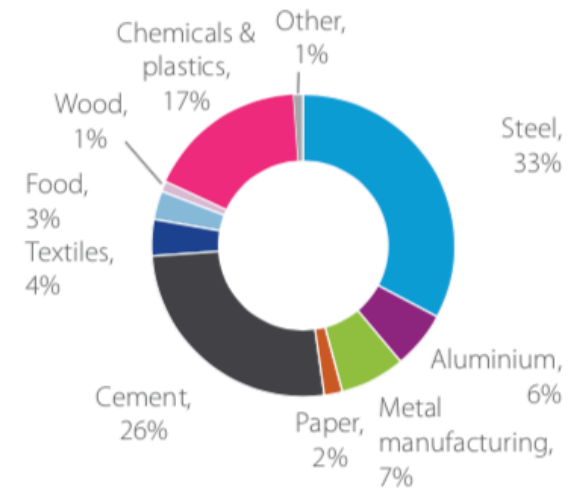


Figure 2.4—Sources of Chinese CO₂ emissions

5 Materials – Breakdown of Production, Energy Intensity, and Carbon Intensity

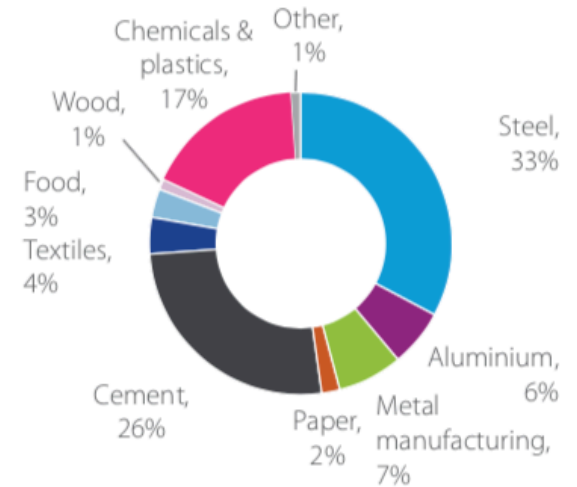


Figure 2.4—Sources of Chinese CO₂ emissions

Material	Global annual production (Mt)	Energy intensity (GJ/t)	Carbon intensity (t CO ₂ /t)
Cement	2,800	5	1
Steel	1,400	35	3
Plastic	230	80	3
Paper	390	20	1
Aluminium	70	170	10

What do we use these materials for?

Conversion device switching + renewable electricity would aid in decarbonizing industry

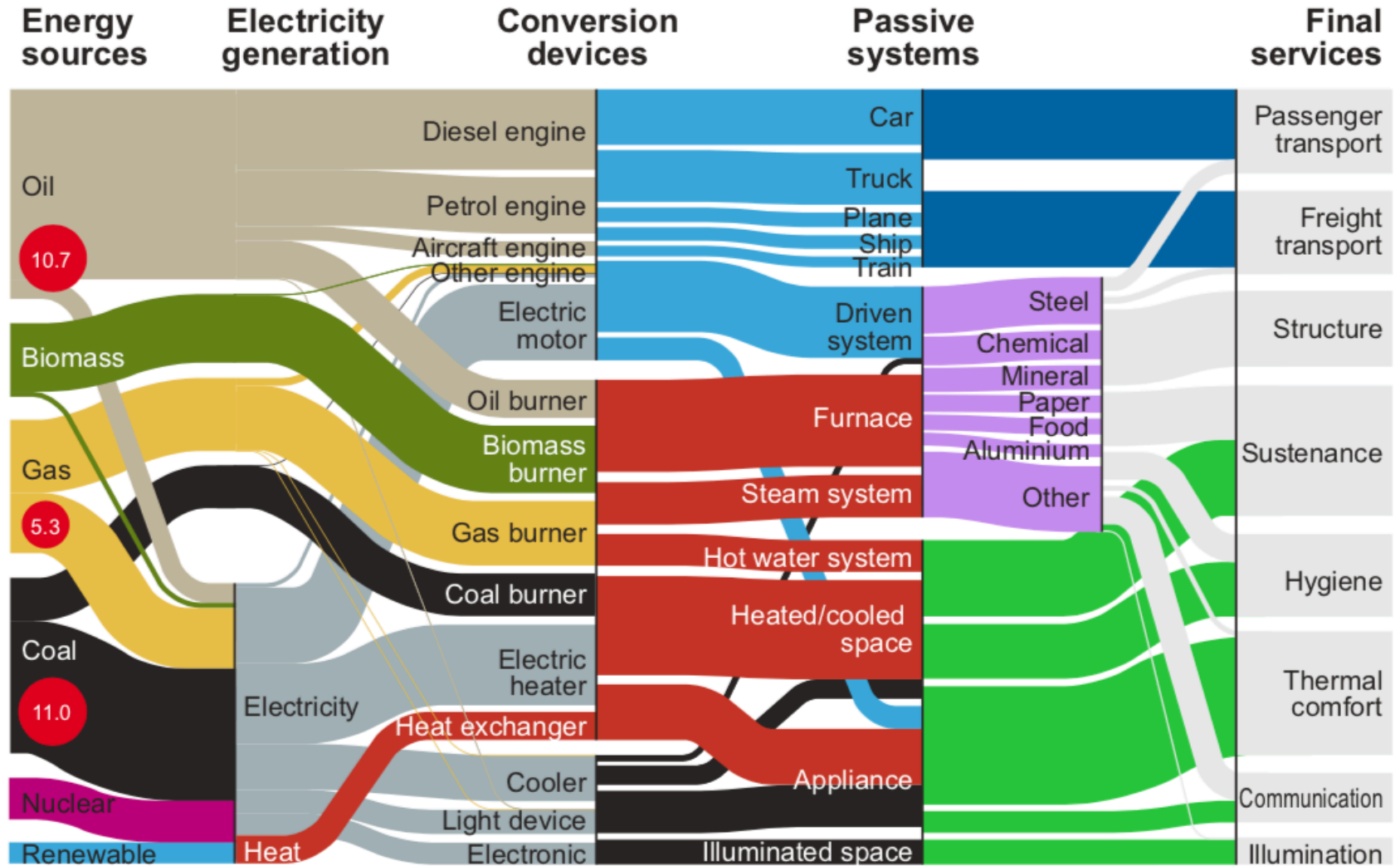
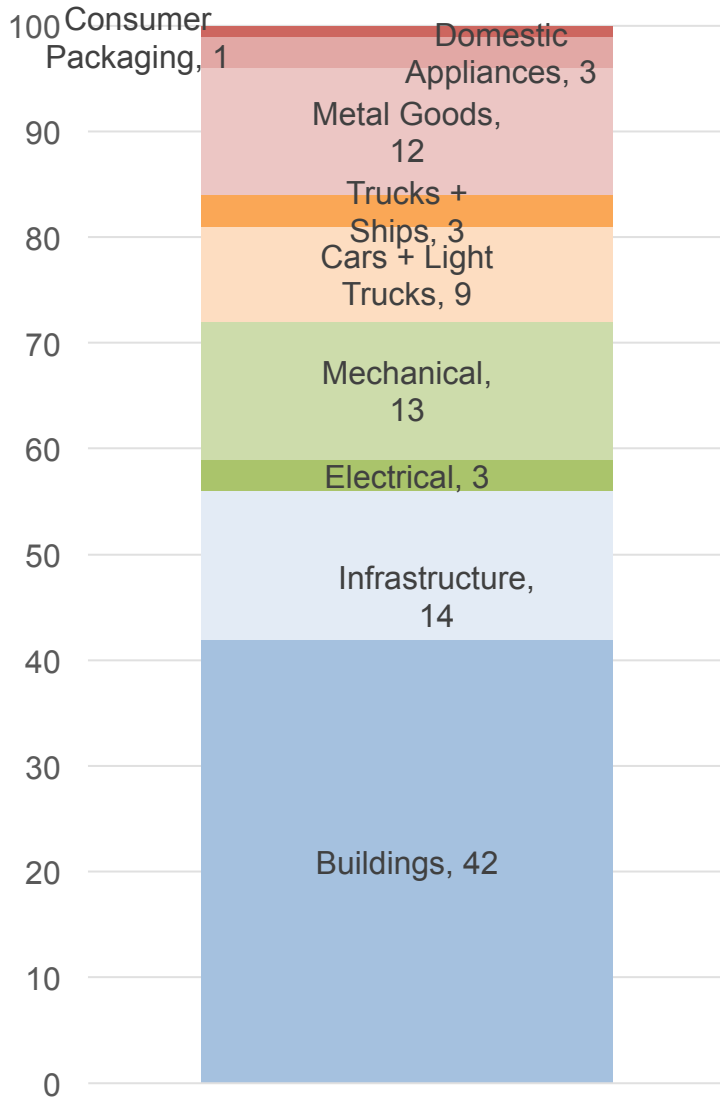


Figure 2.5—Sankey diagram of global energy use¹⁸

Global energy demand in 2005, total = 475 EJ

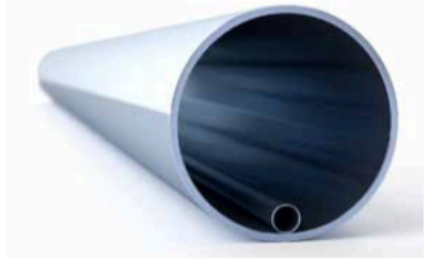
● Global carbon emissions in 2005, total = 27 Gt CO₂

Our uses of steel

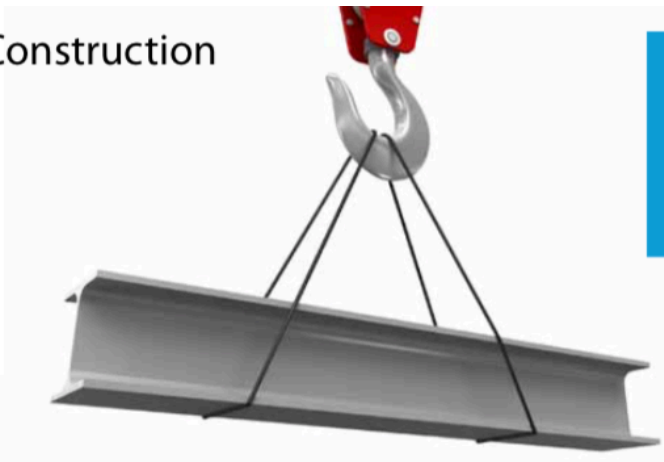


Infrastructure
150Mt
14%

For infrastructure: 24% of steel is in structural sections; 54% is reinforcing bars; 6% is hot rolled train rails (providing a strong, wear and fatigue resistant contact surface); 16% is in pipes formed by welding rolled steel, with high corrosion and fatigue resistance, and high strength to resist internal pressure and installation stresses.



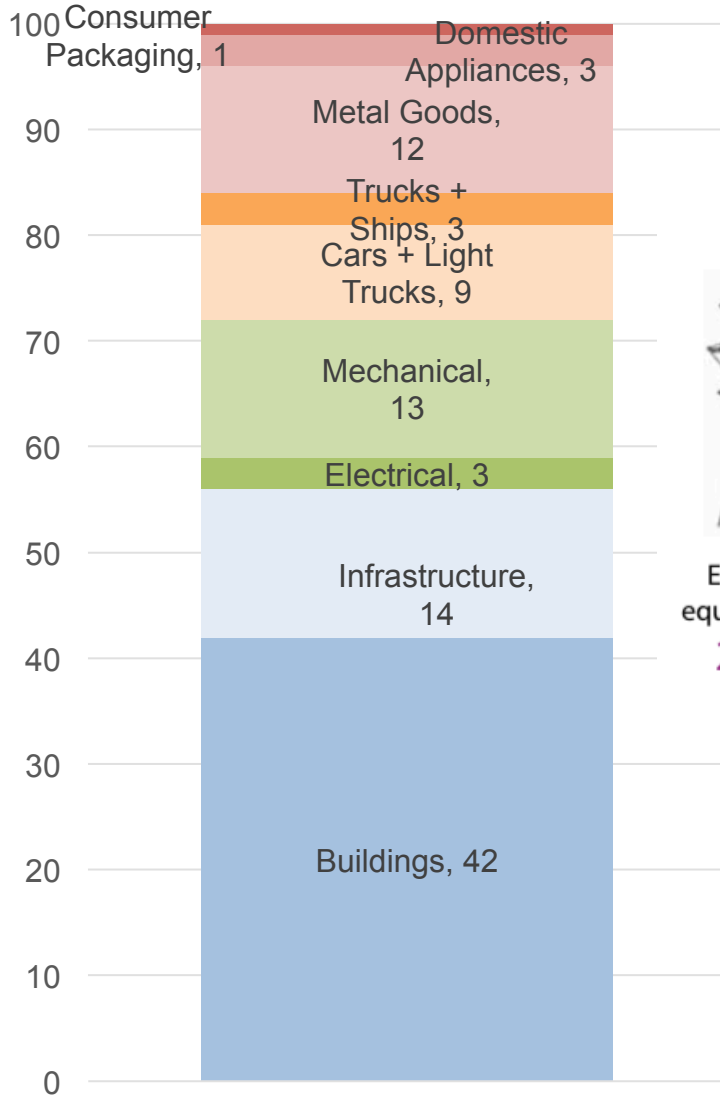
Construction



Buildings
433Mt
42%

25% of the steel in buildings is in structural sections, mainly hot rolled sections but also some welded plate. Sections form a strong, stiff structural frame. 44% is in reinforcing bars, adding tensile strength and stiffness to concrete. Steel is used because it binds well to concrete, has a similar thermal expansion coefficient and is strong and relatively cheap. 31% is in sheet products such as cold-formed purlins for portal frame buildings and as exterior cladding.

Our uses of steel



Electrical equipment
27 Mt
3%

Industrial equipment

30% of steel in electrical equipment is high silicon content electrical steel forming the cores of transformers or the stator and rotor parts of electrical motors. Other major uses include pylons (constructed from bolted, cold-formed, galvanized L-sections forming a light-weight durable tower); and steel reinforced cables (where wound galvanized steel wires provide the strength to carry conducting aluminium in long span transmission cables).

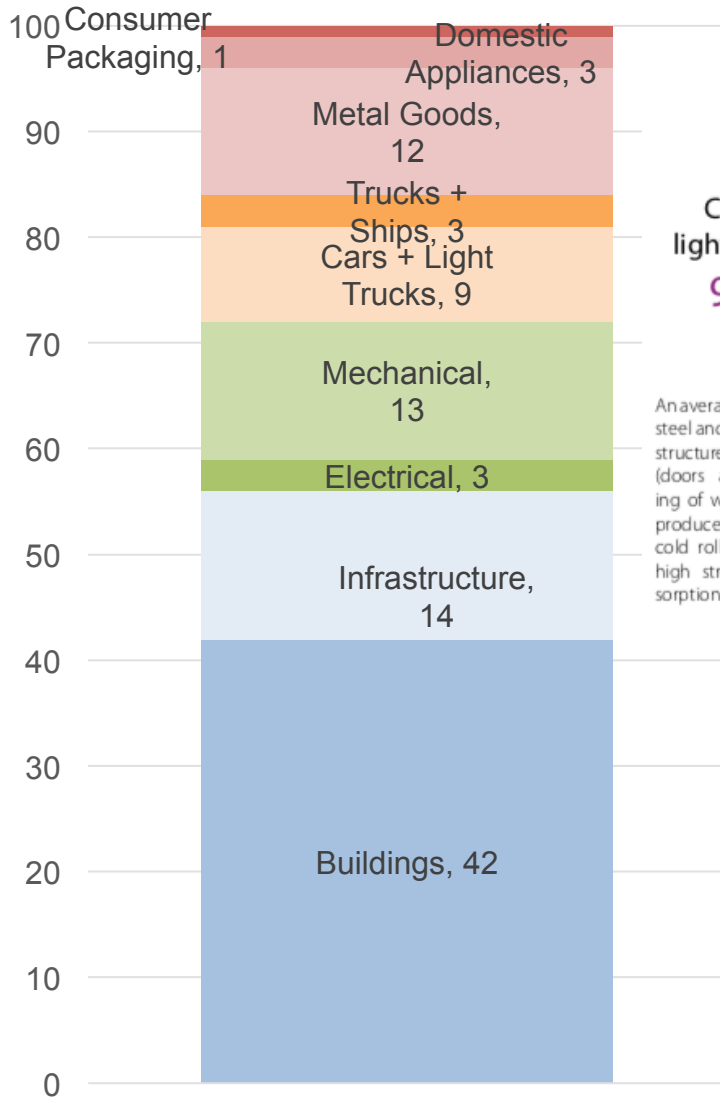
Mechanical equipment
137 Mt
13%



This covers a wide range of equipment from small workshop

rolled bar; tubes contribute a further 22%, as do hot and cold rolled coils. Cast products and wire rod contribute the remainder. 40% of the steel is plate or hot

Our uses of steel



Transport

Cars and light trucks
93 Mt
9%



An average car contains 960 kg of steel and iron. 34% is in the body structure, panels and closures (doors and bonnets), consisting of welded, profiled sections produced by stamping formable cold rolled sheet. This provides high strength and energy absorption in case of a crash. 23%

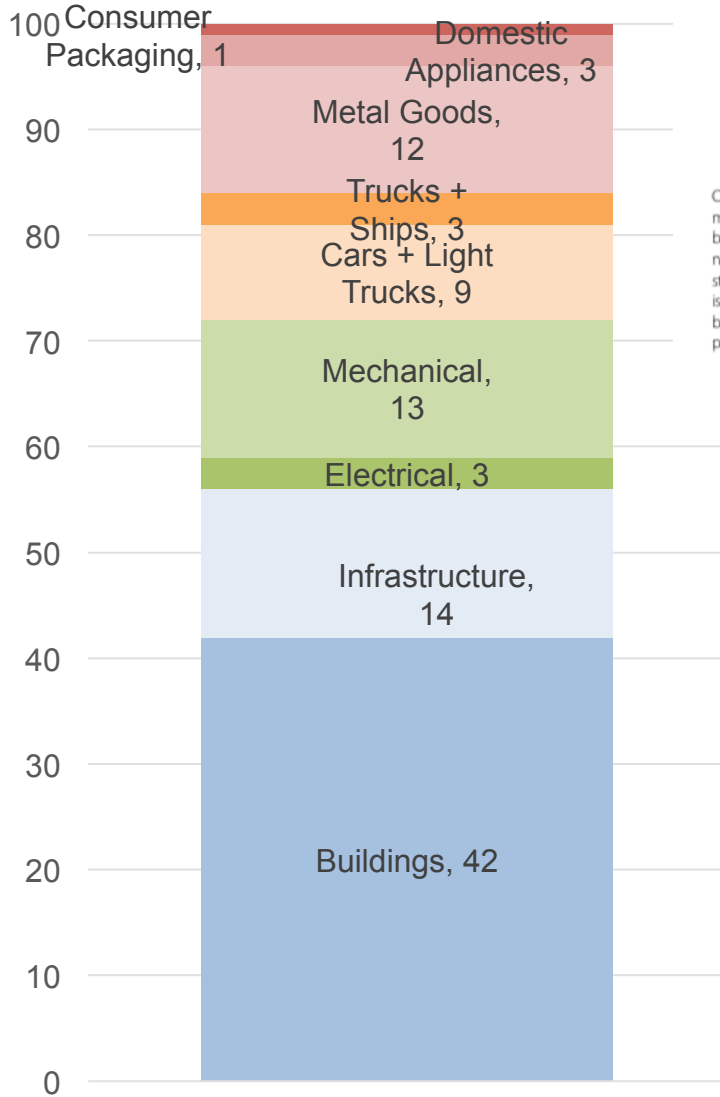
is in the drive train, consisting of grey cast iron for the engine block and machinable carbon steel for the wear resistant gears. 12% is in the suspension, using rolled high strength steel strip. The rest is spread between the wheels, tyres, fuel tank, steering and braking systems.

Trucks and ships
28 Mt
3%



The basic steel components described for the car also apply to trucks, but unlike cars, all truck engine blocks are steel. Frame rails and cross members are usually high tensile steel, and the cab structure and outer skin is often made from galvanized steel. Steel for the ship hull is rolled primary mild steel, providing strong, tough, dimensionally consistent plates that are welded together.

Our uses of steel



Metal products

Metal goods
134 Mt
12%

Other metal goods include a multitude of products, from baths and chairs to filing cabinets and barbed wire. 30% of steel entering this product group is hot rolled coil; 20% is hot rolled bar; and the remainder is either plate, narrow strip, or cast iron.



Consumer packaging
9 Mt
1%



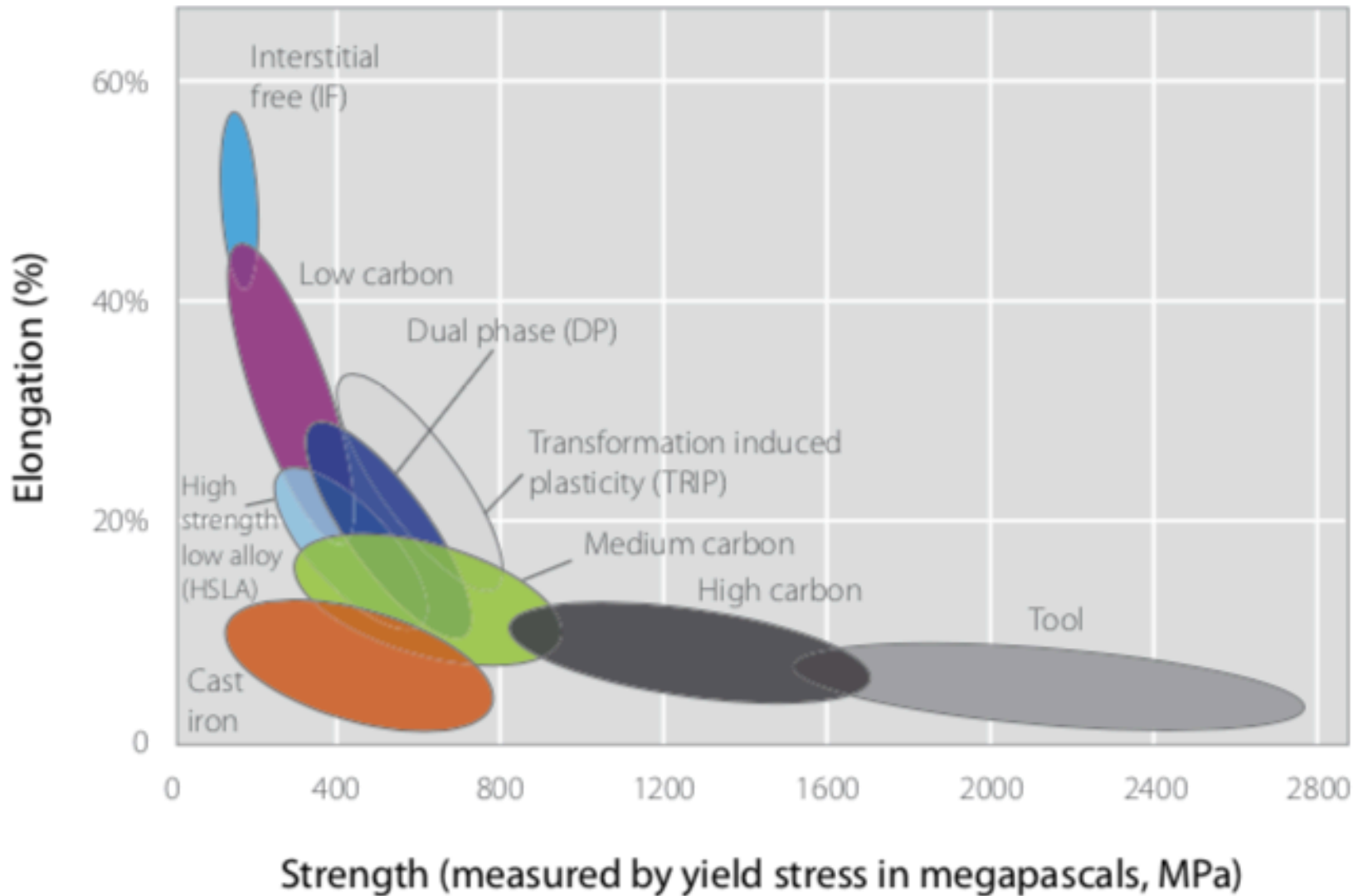
Steel use in packaging is dominated by tin-plated rolled steel, which doesn't corrode. 60% of this steel is made into food cans, providing durable packaging for the subsequent cooking and distribution. 40% is used for aerosols.

Domestic appliances
29 Mt
3%

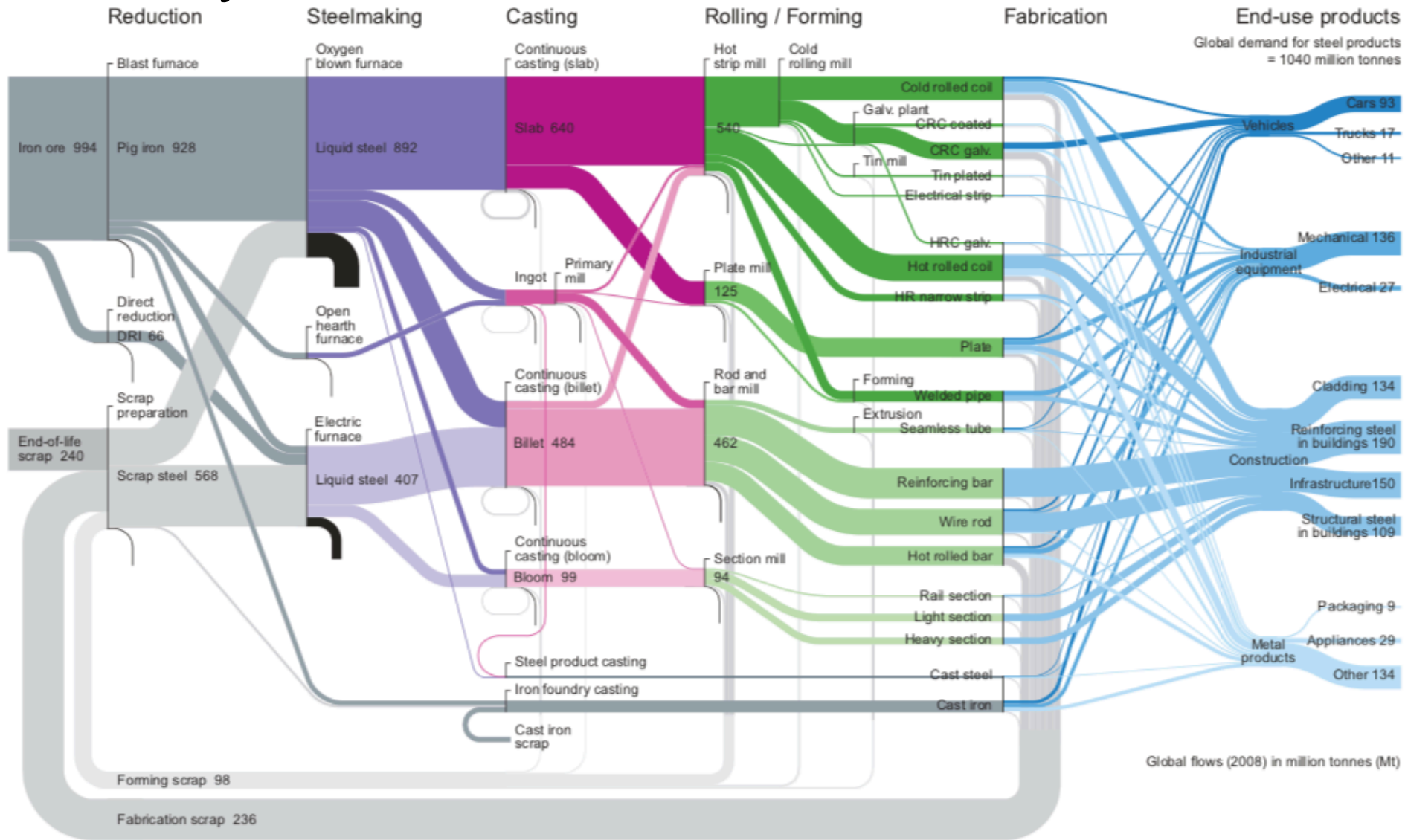


Appliances are dominated by white goods (up to 70%). The vast majority of steel used here is cold rolled coil, often galvanized or painted. Most of this steel is used for panelling. Other applications including washing machine tubs (welded rolled steel strip), motors, expanders in fridge/freezers and cast parts for transmissions.

Typical properties for groups of steels



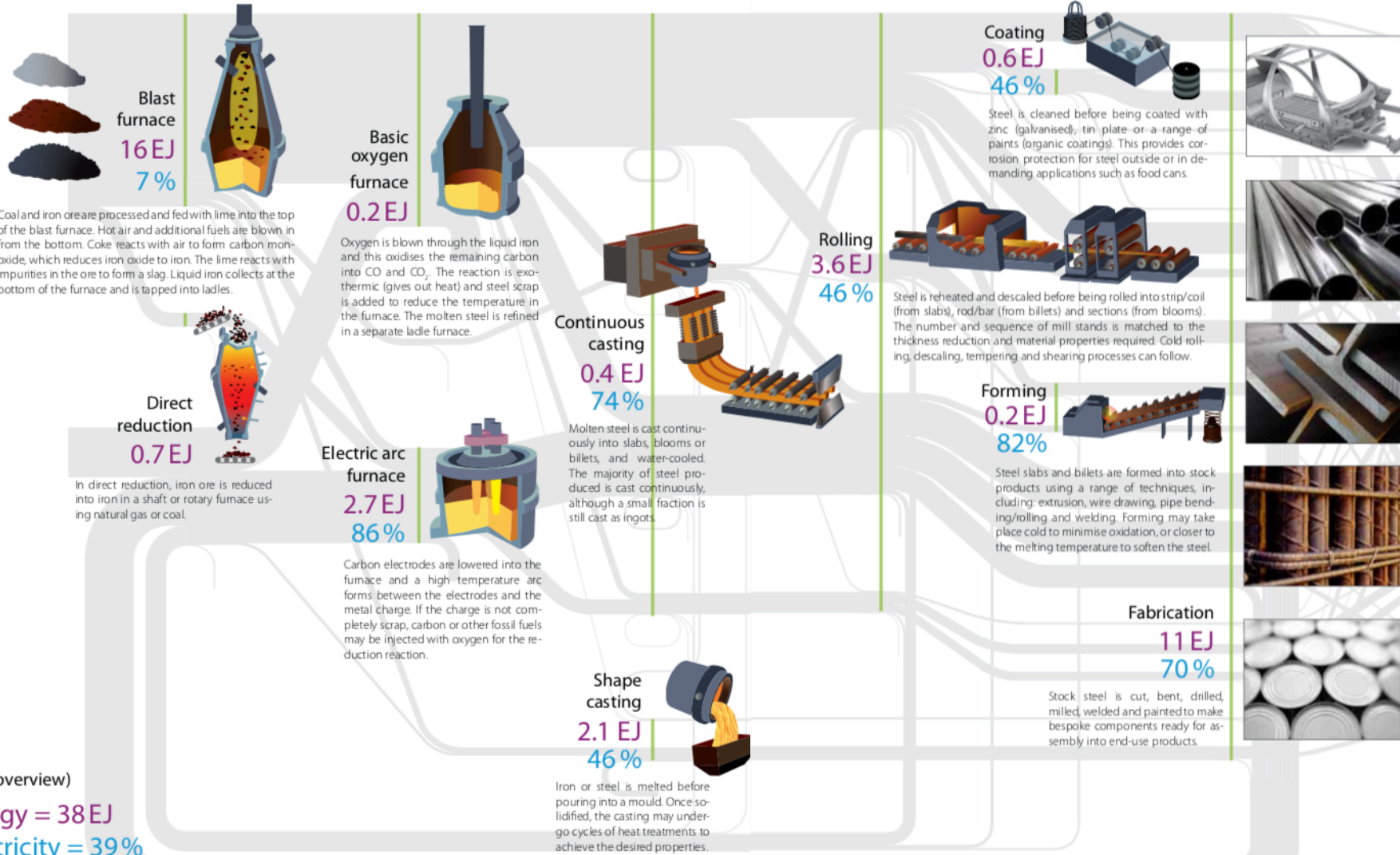
Sankey of Steel Flow



<https://www.youtube.com/watch?v=9l7JqonyoKA>

Most energy used in blast furnace + fabrication

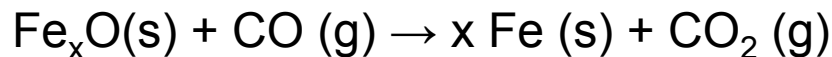
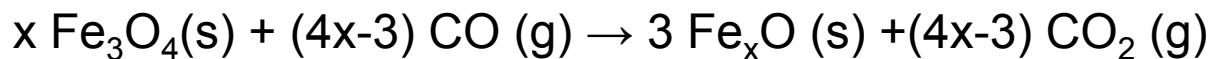
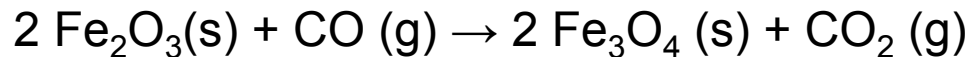
Steel process map



Blast furnace and direct reduction chemistry present opportunities for decarbonization

These numbers depend on chemical production of CO₂

Direct Reduction:



If electricity were greener, these numbers would be lower

Process	Emissions (t CO ₂ /t)
Iron making—blast furnace	0.5
Coking	0.2
Sintering	0.4
Direct-reduction	1.2
Steelmaking—oxygen blown furnace	0.2
Steelmaking—electric arc furnace	0.5
Scrap preparation	0.01
Steelmaking—open hearth furnace	1
Continuous casting	0.01
Ingot casting	0.05
Hot strip mill	0.1
Cold strip mill	0.4
Plate mill	0.1
Rod and bar mill	0.2
Section mill	0.2
Galvanising plant	0.2
Tinning mill	0.04
Extrusion	0.2
Primary mill	0.1
Forming	0.1
Steel product casting	2.4
Iron foundry casting	1.7
Fabrication	1

Table 5.1—Emissions estimates per unit processed for major steel production processes⁵

Novel processes for steel making

- Options for using less coke
 1. Substitution of other carbon fuels
 2. Hydrogen as a fuel
 3. Top gas recycling
- Alternative ways to make iron
 1. Direct reduced iron
 2. Smelt reduction
 3. Electrolysis

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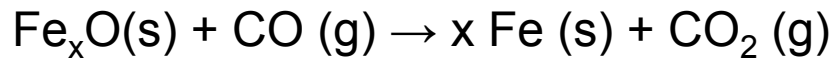
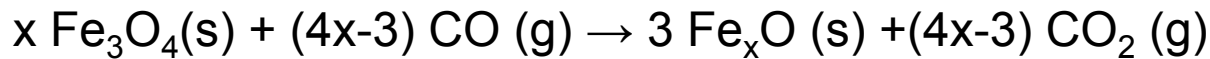
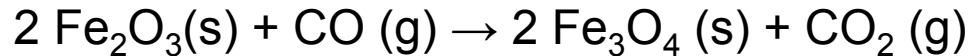
Substitution of other carbon fuels

- Substitute coke by biomass, charcoal, waste plastics.
- Expansion in use limited by rate we can harvest biomass
 - If we replaced all coke in 30 Mt of steel consumption (what the UK consumes), the UK would need to use nearly half its land surface area for charcoal production.
 - Biomass substitution technically feasible, not really “sustainable”



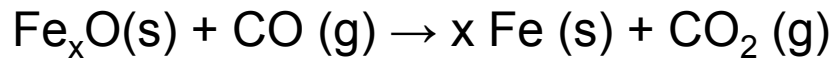
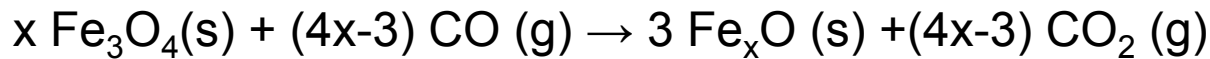
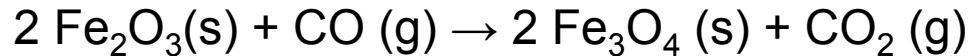
Could H₂ be used as a reductant instead?

Direct Reduction:



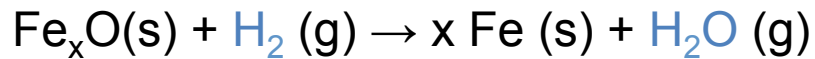
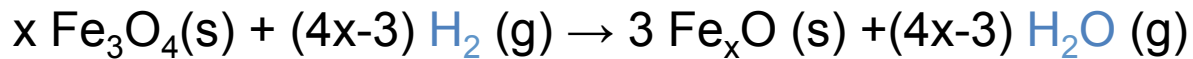
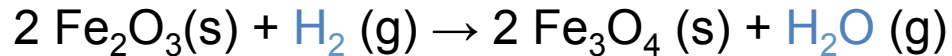
Could H₂ be used as a reductant instead?

Hydrogen-based Reduction:



Could H₂ be used as a reductant instead?

Hydrogen-based Reduction:



Could H₂ be used as a reductant instead?

Hydrogen-based steelmaking is believed to be technically sound.
But...

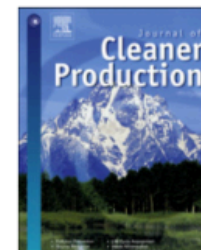
1. Overall emissions impact depends how H₂ is produced – no sufficiently-sized C-free source (yet)
2. Overall emissions impact depends how blast furnace is heated to operating temperature (900-1300 °C)
3. Large-scale adoption of hydrogen substitution looks extremely unlikely, even by 2050.
4. Hydrogen embrittlement



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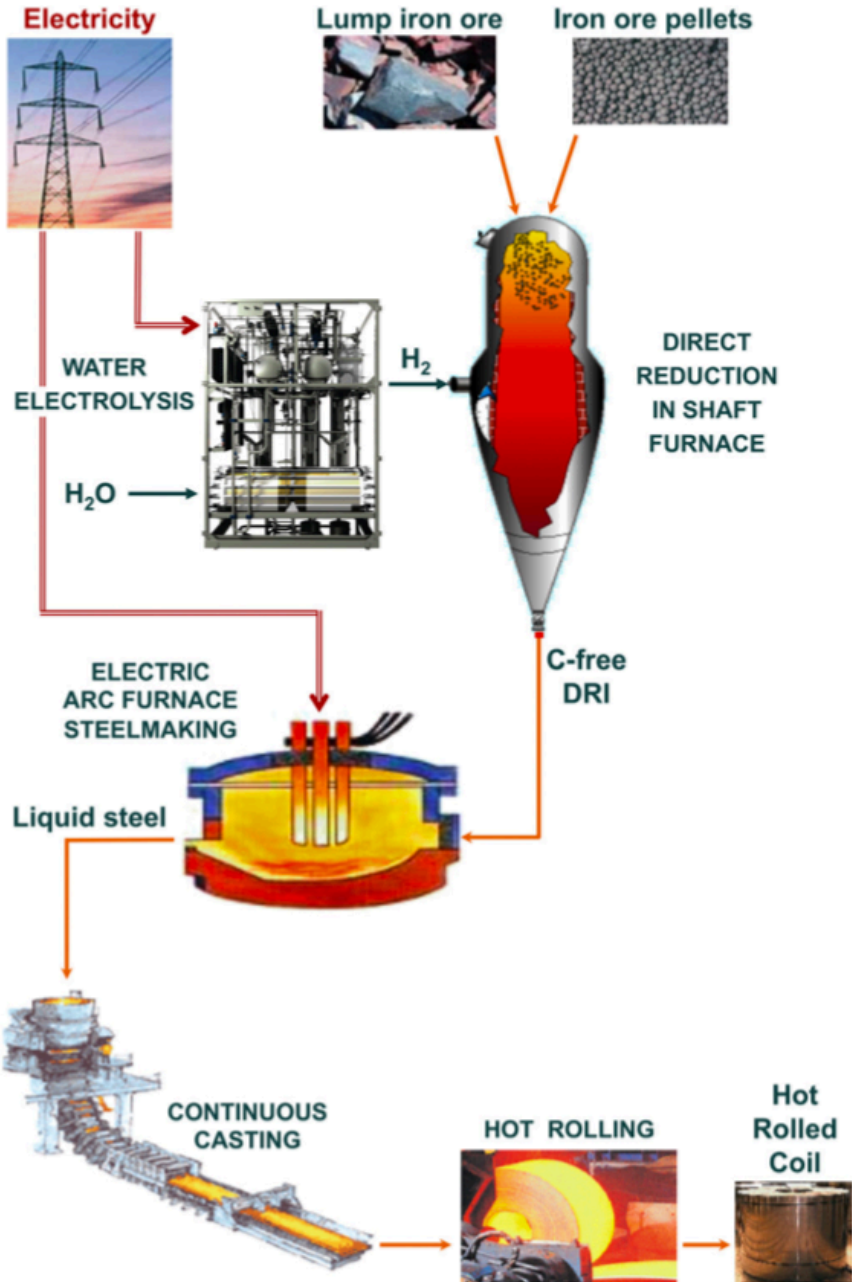
Modelling a new, low CO₂ emissions, hydrogen steelmaking process

A. Ranzani da Costa, D. Wagner¹, F. Patisson*

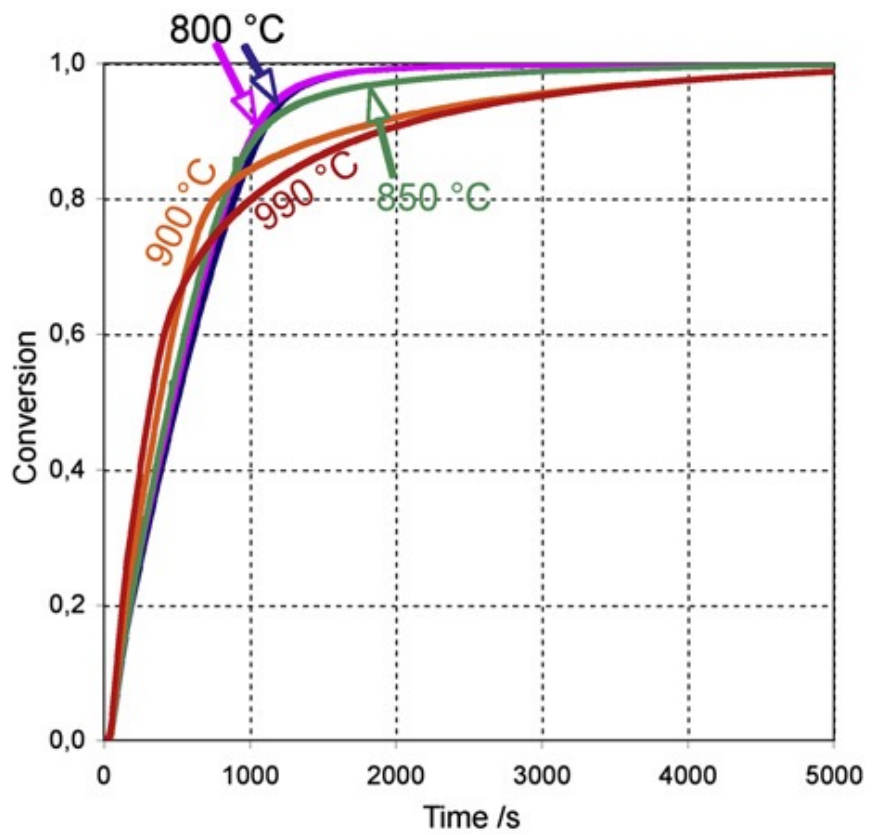
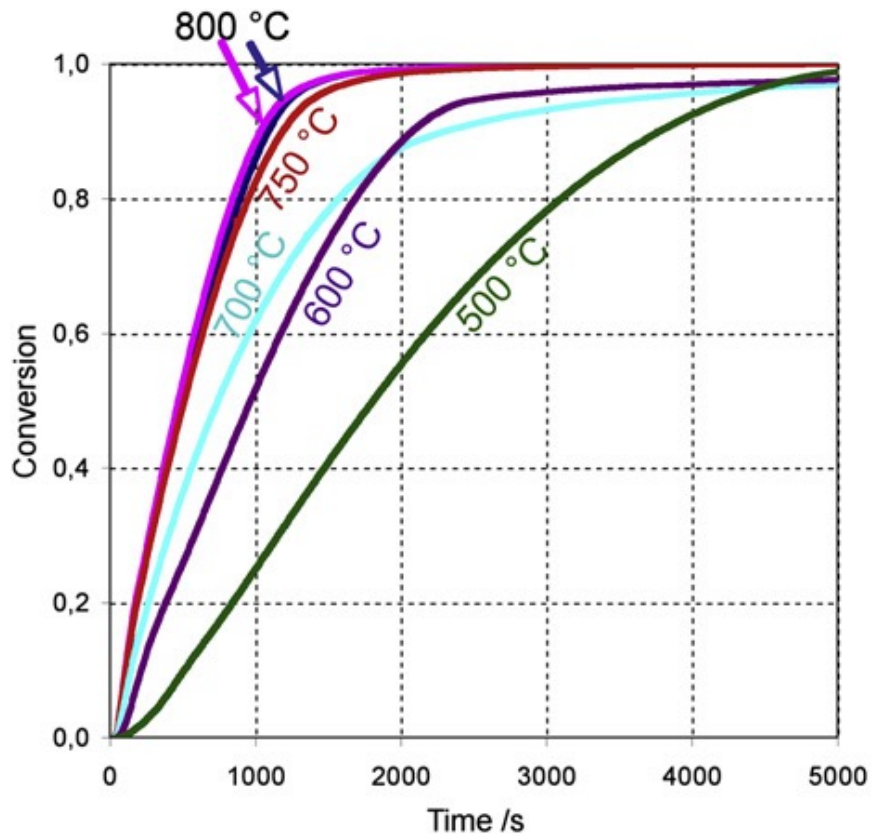
Institut Jean Lamour, Labex DAMAS, CNRS, Université de Lorraine, Parc de Saurupt, 54042 Nancy, France



One potential C-free source is water electrolysis from renewable electricity

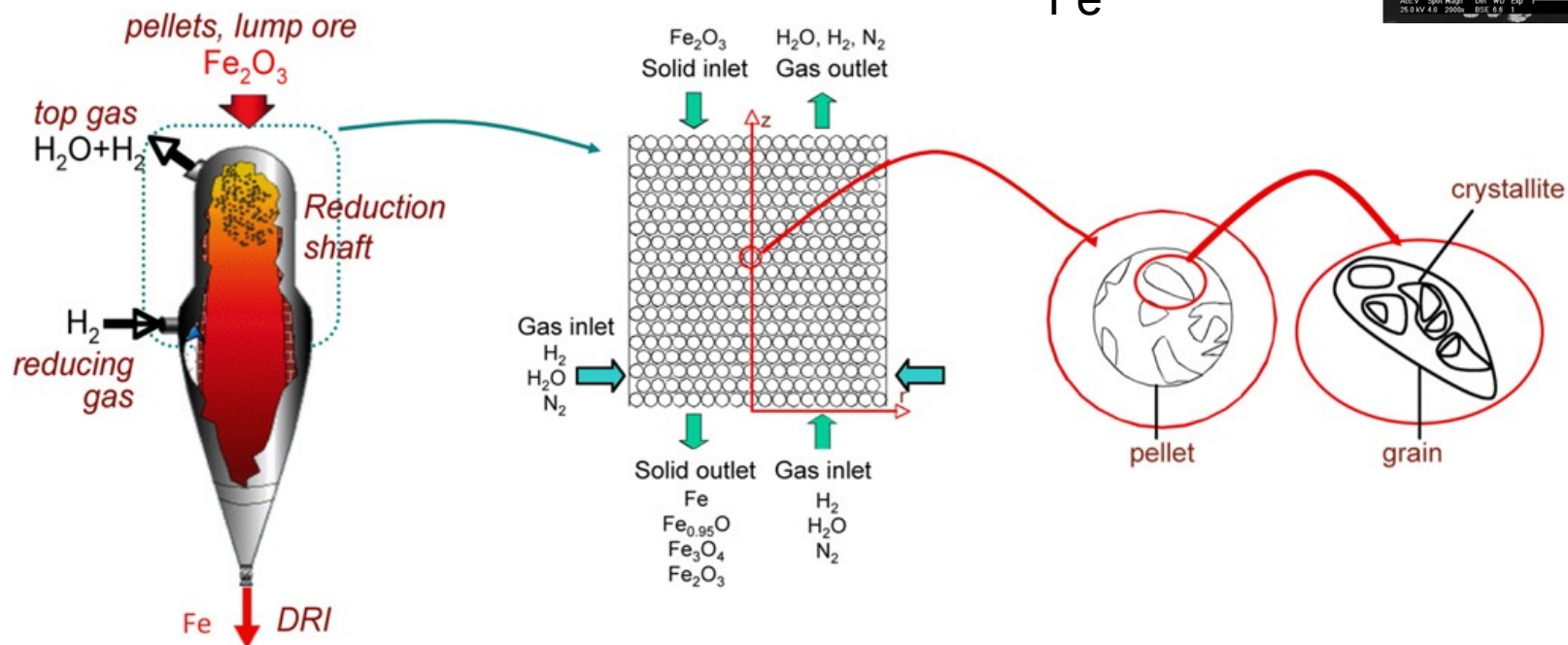
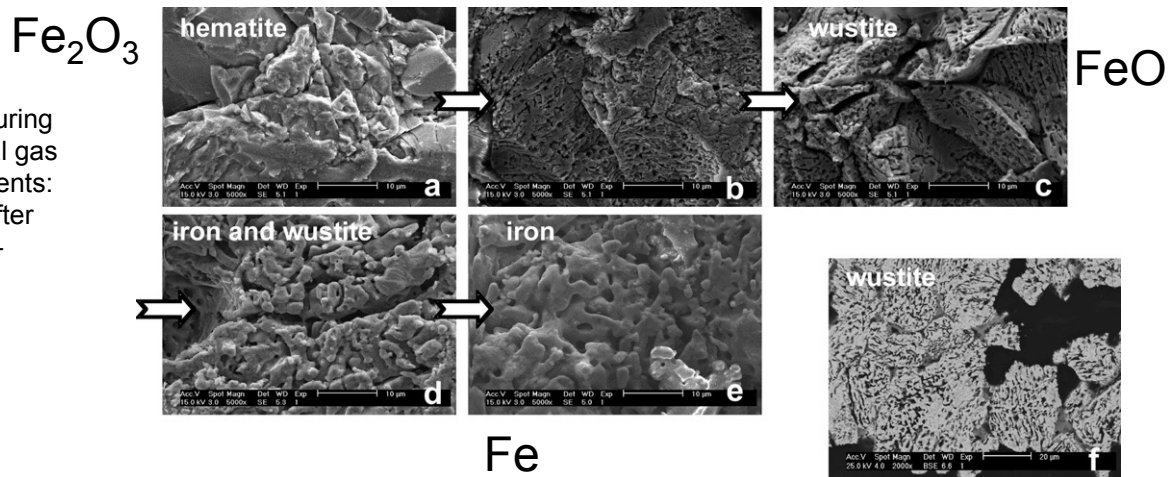


TGA Experiments indicate an optimal temperature of 800 °C for reaction of $\text{H}_2 + \text{Hematite} \rightarrow \text{Fe} + \text{H}_2\text{O}$



A kinetic sub-model was built to simulate the reduction of a single pellet by H₂

Fig. 4. Microstructure evolution of the solids during reduction at 800 °C by H₂ (200 mL /min of total gas flow, H₂/He 60/40 %vol.). Interrupted experiments: (a) initial; (b) after 89 s; (c) after 206 s; (d) after 400 s; (e) fully reduced; (f): after 400 s, cross-section of polished sample.



Model Results

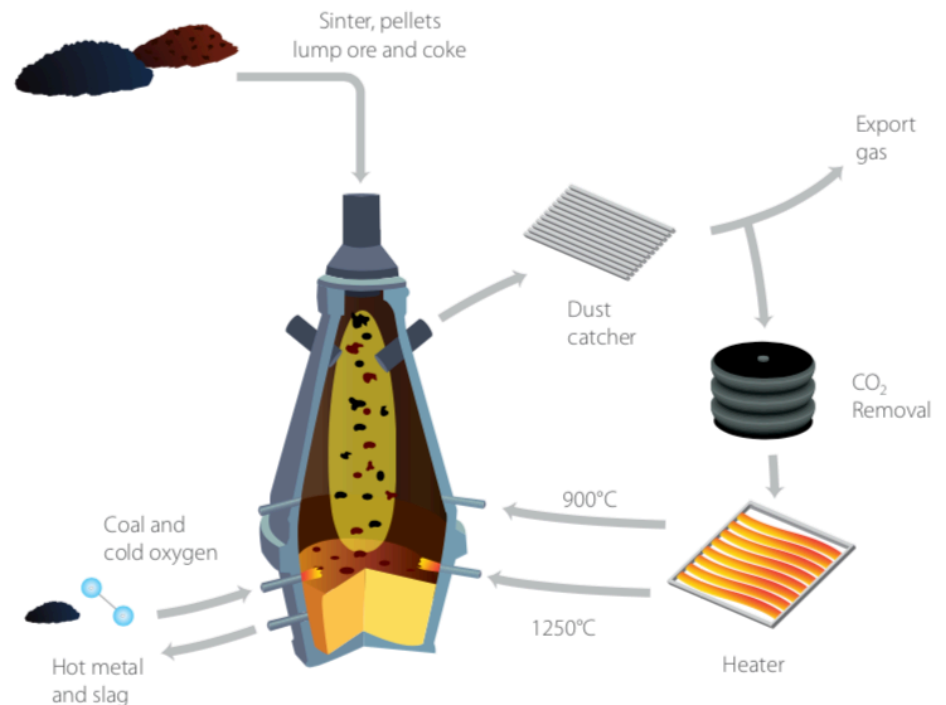
- Complete conversion to metallic iron using pure H₂ could be obtained using a more compact reactor than current industrial DR furnaces
- Reduction by H₂ is faster than that by CO.
- Size of the pellets and the temperature of the inlet gas have a strong influence on the reduction rate: the smaller the pellet diameter, the faster the reduction and thus the more compact the reactor.
- The optimum temperature was 800 °C.
- At higher temperatures, the densification of iron due to sintering causes the reaction to slow down at high conversions.
- The next steps of this work should be to verify if the kinetics used are valid for entire pellets and to adapt the model for the reduction of iron ore by H₂/CO mixtures to validate the model against operation data of existing industrial DR process.

Novel processes for steel making

- Options for using less coke
 1. Substitution of other carbon fuels
 2. Hydrogen as a fuel
 3. Top gas recycling

Top gas recycling

- Blast furnace off gases include CO and CO₂.
- Separate CO from gas stream, use it again as reductant (instead of new C, coke)
- Would want to use O₂ instead of air, because N₂ / CO hard to separate.
- Reduces CO₂ emissions from blast furnace by 5 – 10%.
- Could be coupled with CCS if O₂ is used.



Novel processes for steel making

- Options for using less coke
 1. Substitution of other carbon fuels
 2. Hydrogen as a fuel
 3. Top gas recycling
- **Alternative ways to make iron**
 1. Direct reduced iron
 2. Smelt reduction
 3. Electrolysis

Direct reduced iron (no coke) – widely used in countries with NG supplies

- No coke
- Lower T than blast furnace (800 – 1000 °C)
- High C/impurity concentration, thus
- Refine in EAF
- Most DRI sites use NG to heat, giving off CO₂.

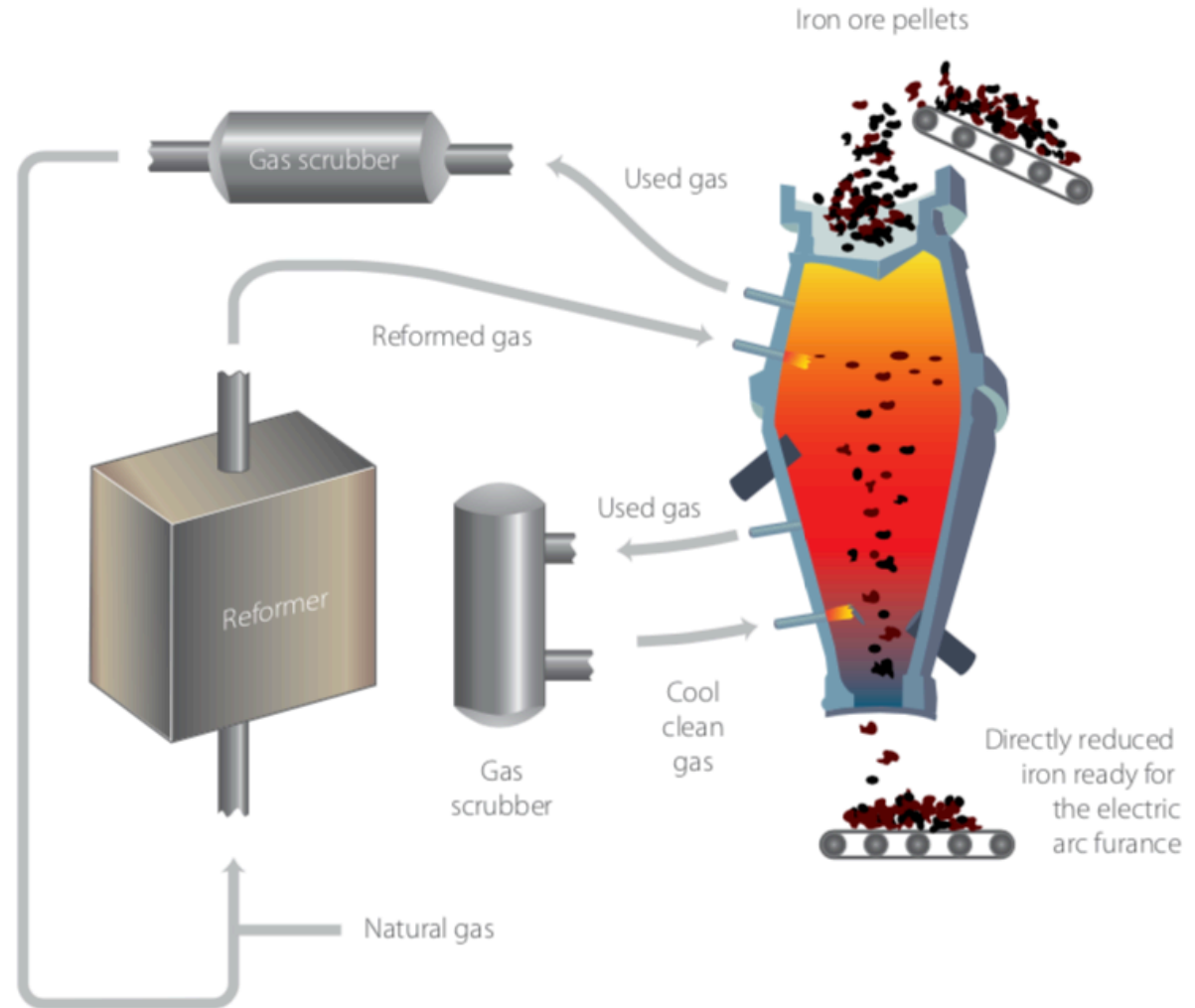


Figure 9.1—Direct reduced iron production¹⁵

Smelt reduction – demonstrations planned (Hisarna pilot plant in The Netherlands)

- No coke
- Lower T than blast furnace (800 – 1000 °C)
- High C/impurity concentration, thus
- Refine in connected pool of **molten** iron; coal and oxygen drive removal of C/impurities

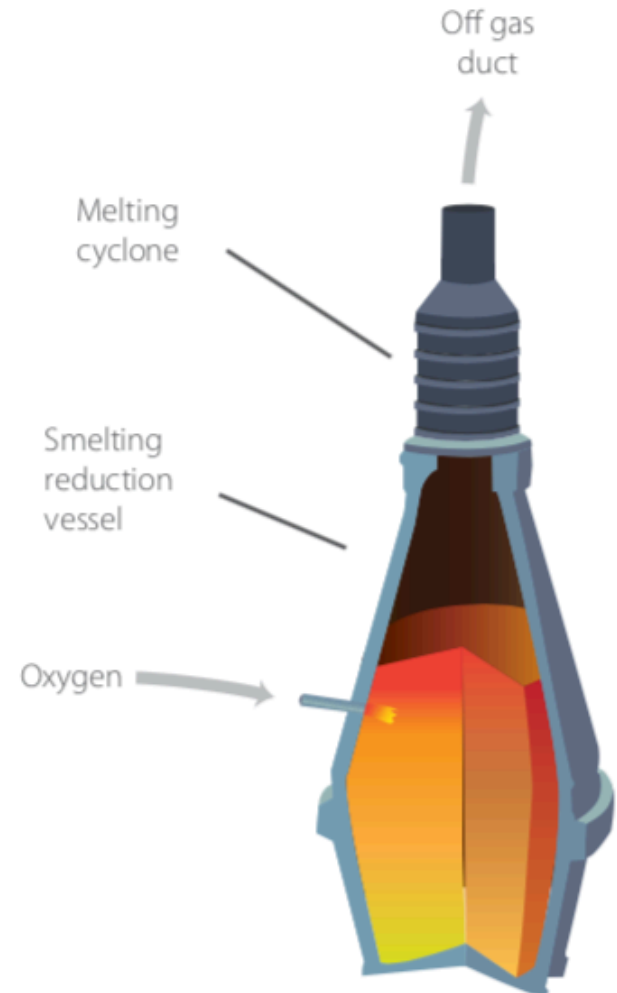
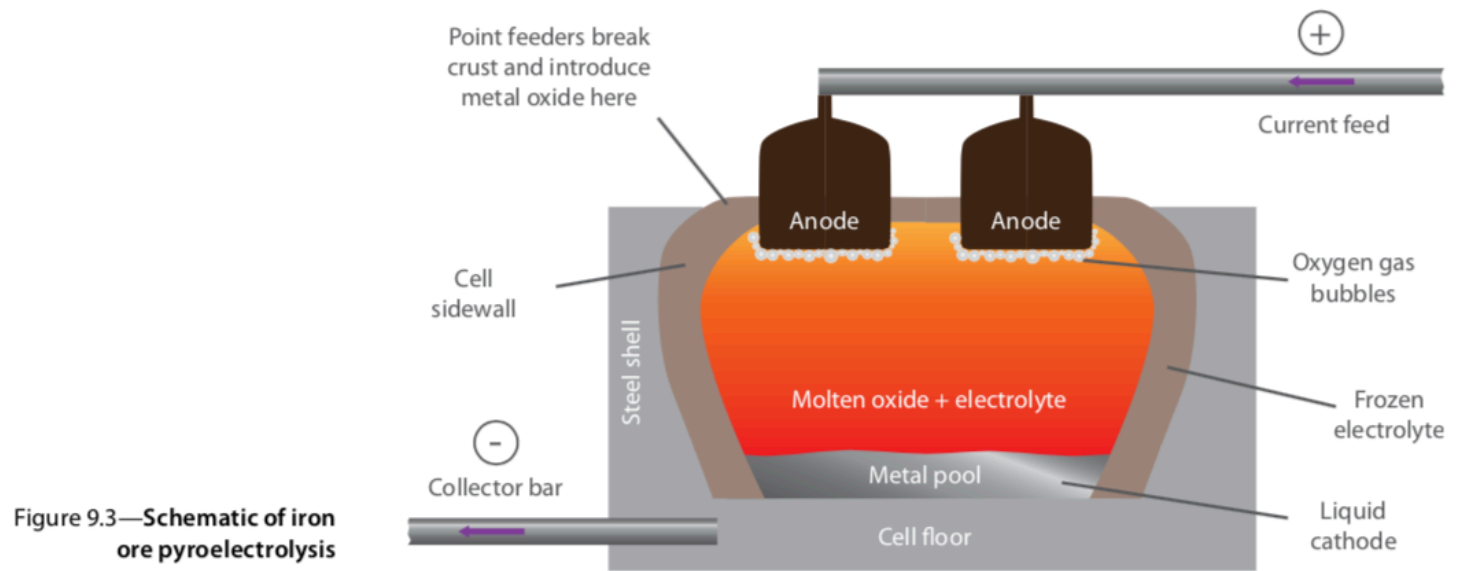


Figure 9.2—Smelt reduction process¹⁵

Electrolysis

- Electrowinning
 - Strong current passed into liquid via positive terminal (an oxide) so pure Fe is electroplated onto negative terminal
- Iron ore pyroelectrolysis
 - Current passed from inert positive terminal through molten iron ore at 1600 °C, forming liquid Fe while O₂ is released.



Sustainable Steel?

Option	CO ₂ abatement potential
Energy efficiency—best available technology	13% for all processes
Direct reduced iron	20% compared to the blast furnace
Smelt reduction	20% compared to the blast furnace
Electrolysis with nuclear power	80% compared to the blast furnace assuming low carbon electricity
Top gas recycling and fuel substitution	10% compared to the blast furnace
Electric motors in fabrication	50% reduction in energy

Table 11.1—Summary of the emissions abatement from energy efficiency and novel technologies for steel



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